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(vi) $\text{HCOOC}(\text{CH}_3)_3$ is an isomer of X. For the ^1H NMR spectrum of this isomer, deduce the total number of peaks (excluding the TMS peak at 0 ppm) and the ratio of peak areas. For each peak, deduce whether it is a singlet, doublet, triplet, quartet or shows a more complex splitting pattern. [3] Number of peaks (excluding the TMS peak at 0 ppm):

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(iii) The graph below indicates the pH change during the titration of 20.0 cm³ of 0.100 mol dm⁻³ of CH₃COOH (aq) with 0.100 mol dm⁻³ KOH (aq). From the graph, identify the volume of KOH (aq) and the pH at the equivalence point. pH
14 13 12 11 10 9 8 7 6 5 4 3 2 1 0 0 10
20 30 40 50 Volume of KOH (aq) added /
cm³ [2]

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